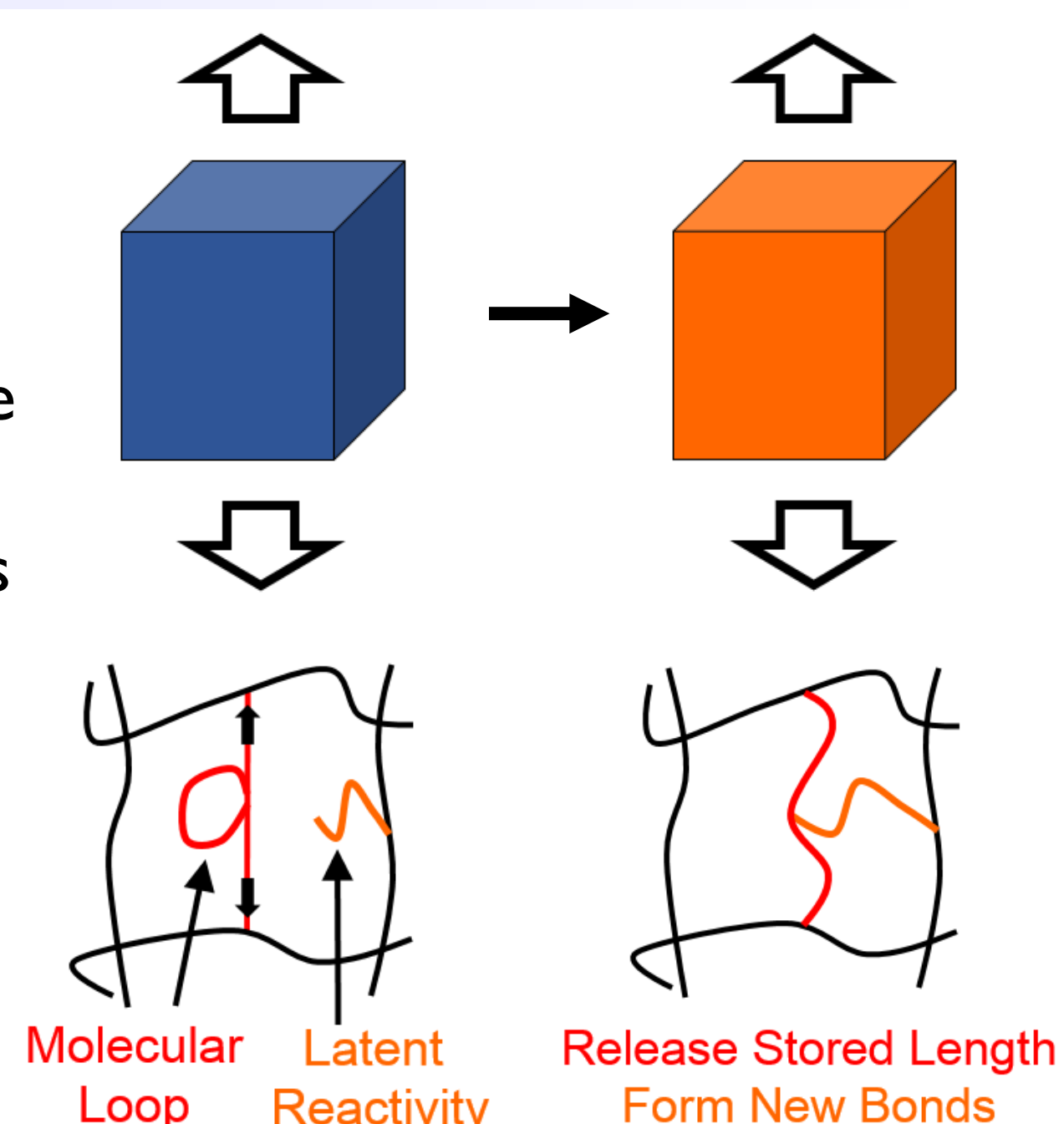


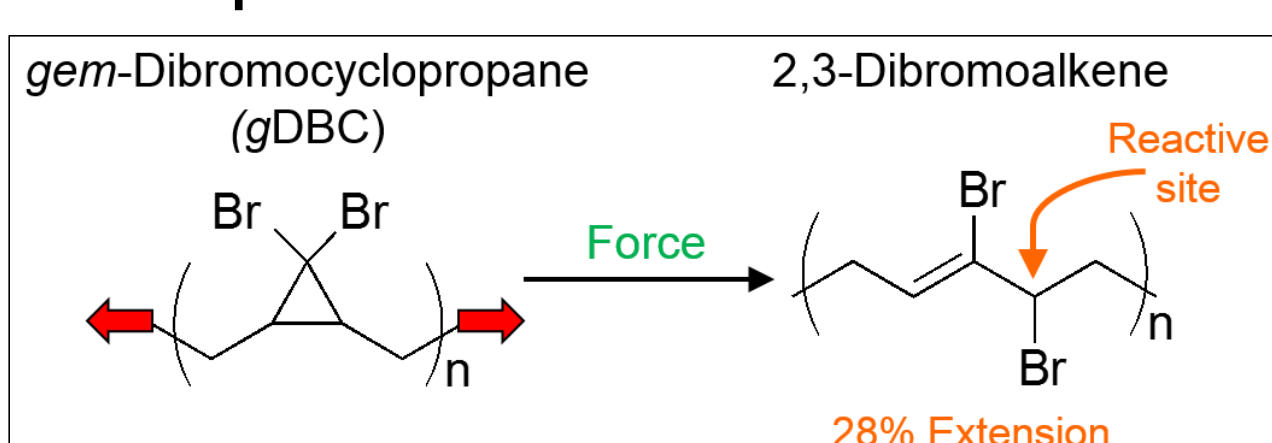
Introduction

A mechanophore is a chemical functional group that is activated by mechanical force.¹

- Produces color change, fluorescence, small molecule release
- Applications in sensors, data displays, self-strengthening materials



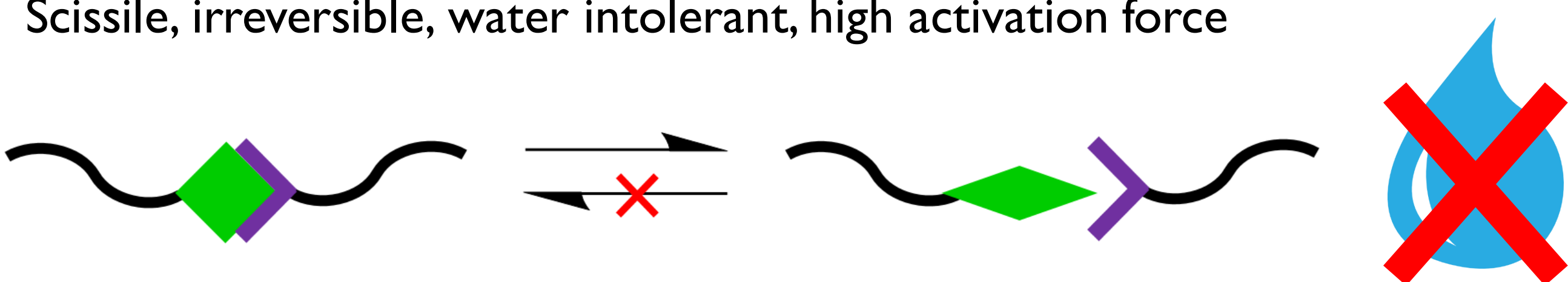
Example:



Project Design

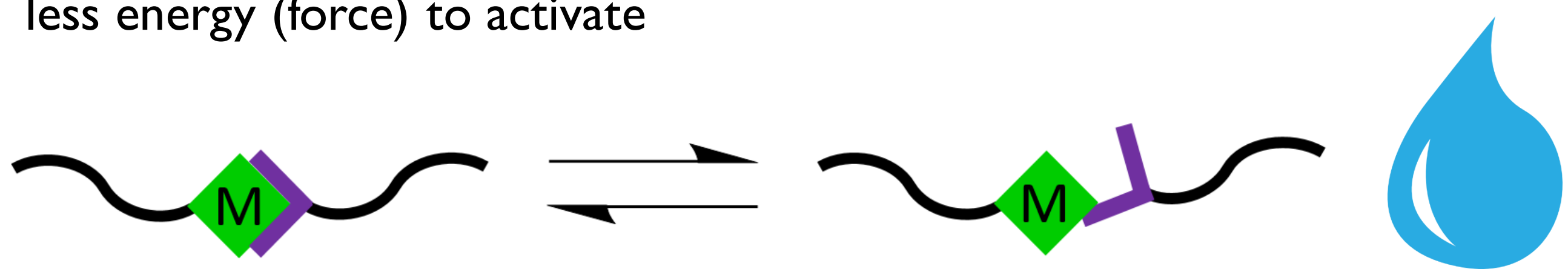
Problems with most current systems:

Scissile, irreversible, water intolerant, high activation force



Organometallic complexes could solve these problems:

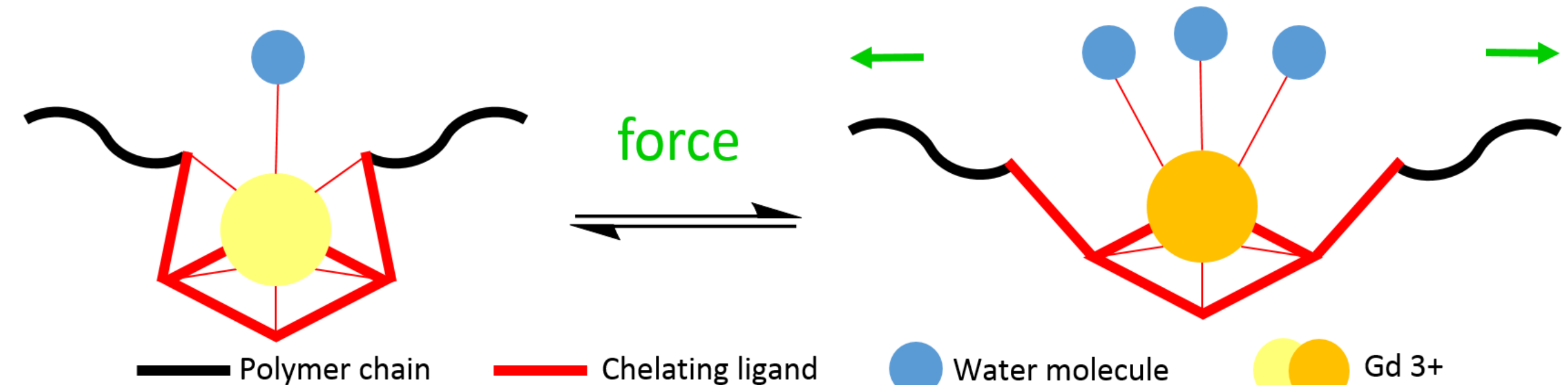
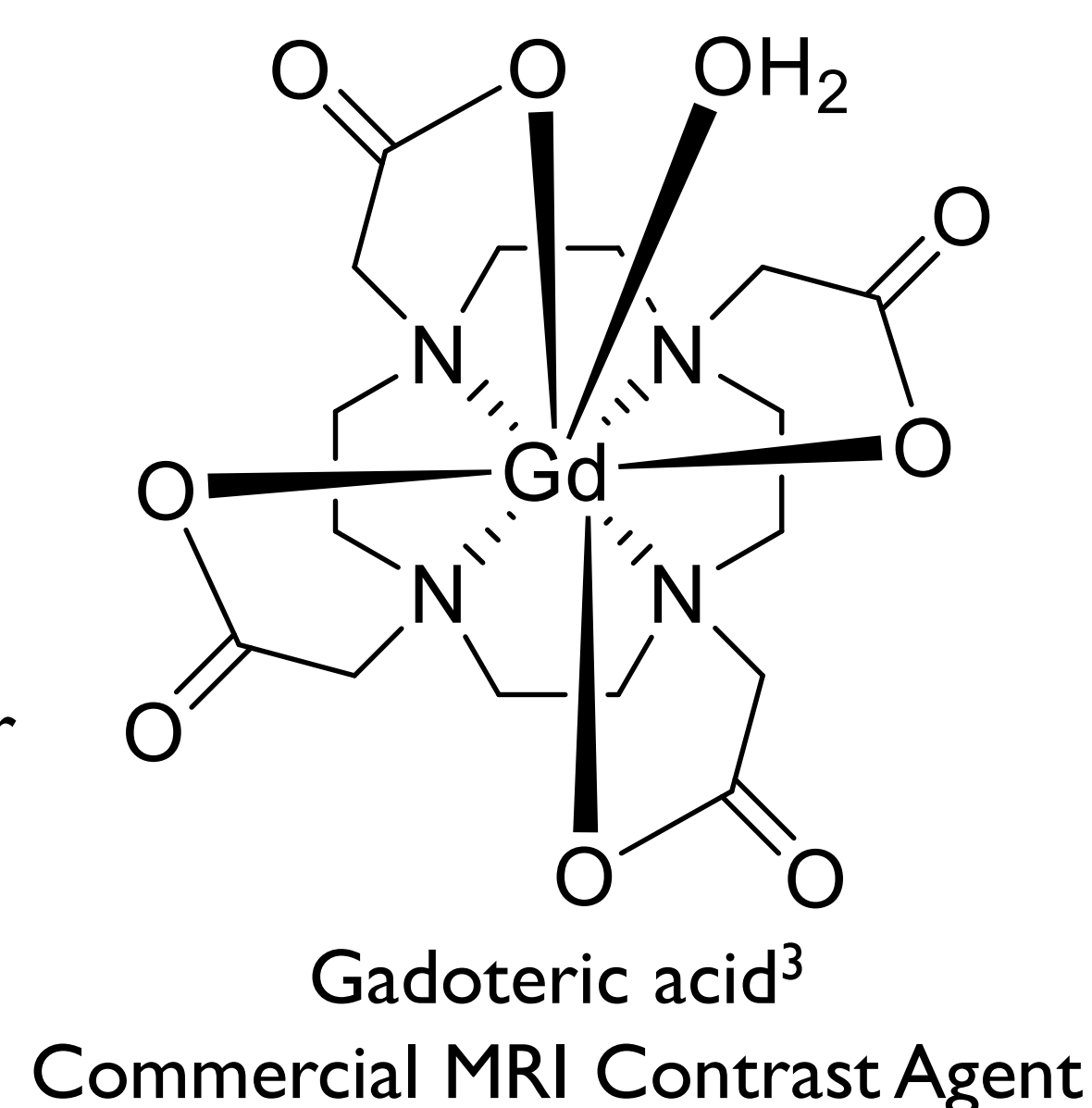
Metal-ligand bonds are weaker than organic covalent bonds, requiring less energy (force) to activate



Magnetic Resonance Imaging (MRI) contrast agents provide an excellent chelate design³ which can be modified to act as a mechanophore.

Strategy:

1. Complex incorporated into polymer backbone
2. Force applied to bulk polymer
3. Force transferred from polymer to complex
4. Parts of ligand dissociate
5. Solvent (water) coordinates in their place
6. Change in coordination sphere = change in color



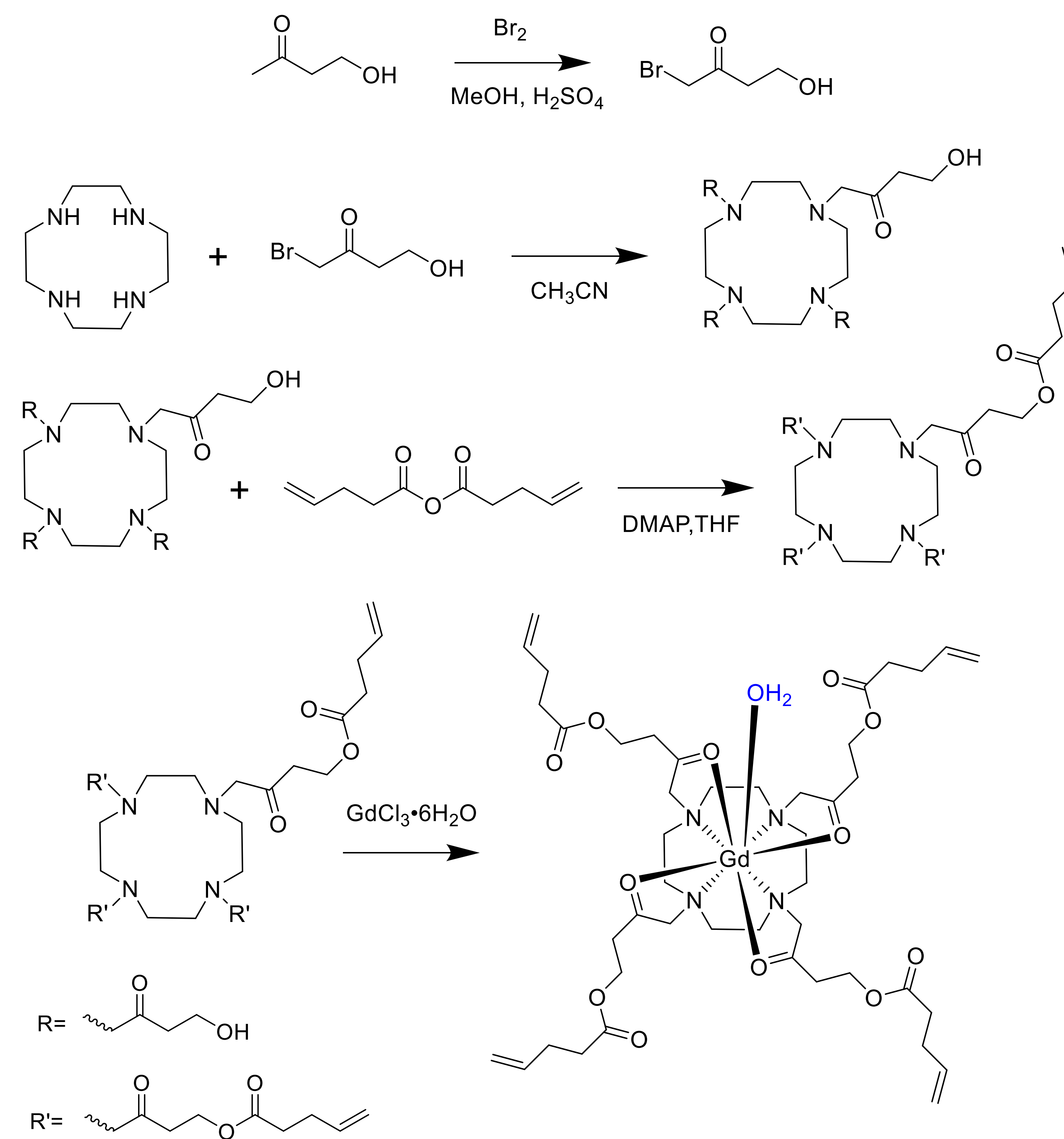
Goals

A) Design a metal complex as an effective mechanophore that changes color under applied force as a colorimetric tension sensing system.

B) Synthesize the target mechanophore and perform mechanical testing.

A successful project would provide proof of concept for a totally new class of mechanophore.

Synthesis



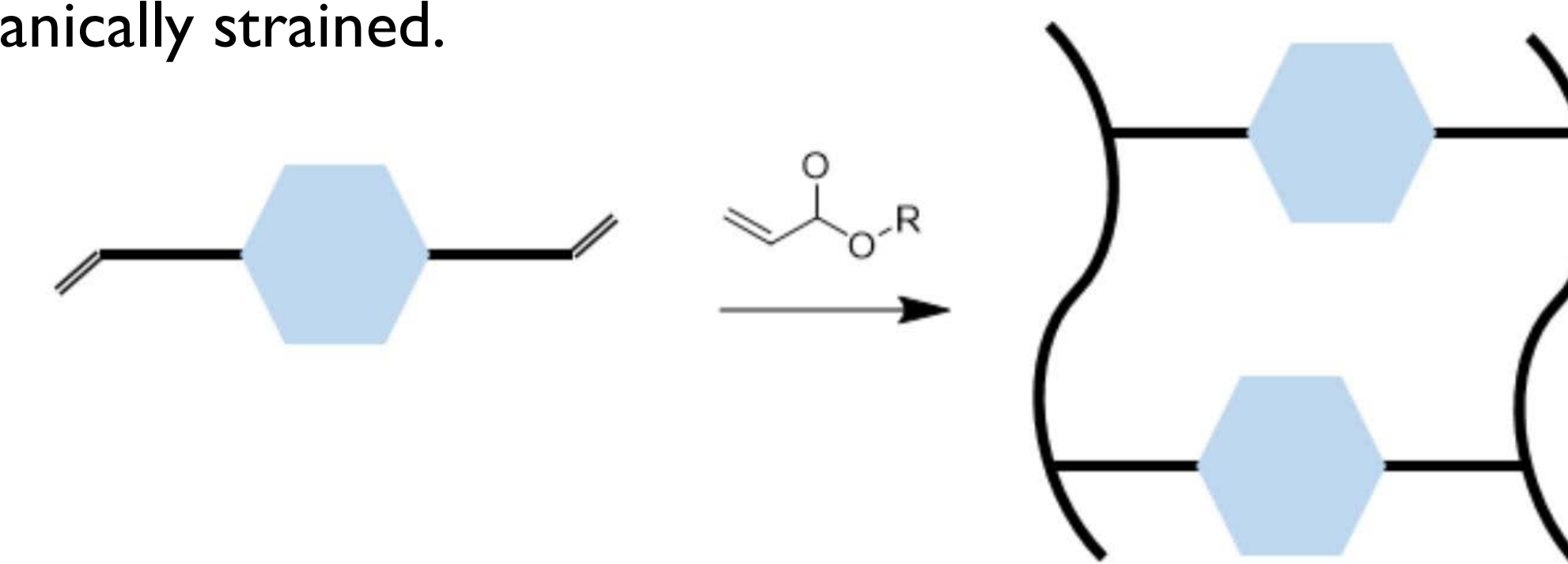
Reaction conditions optimized and monitored by ¹H NMR. Absorbance, excitation and emission data will be collected on the final complex by UV-Vis spectroscopy. Synthesis based off sources 5,6.

References

1. Brown, C. L.; Craig, S. L., *Chemical Science* **2015**, 6 (4), 2158-2165.
2. Ramirez, *Nature Chem.*, 2013, 5, 757
3. Guglielmo, F. F.; Mitchell, D. G.; Gupta, S., *Radiol Clin North Am* **2014**, 52 (4), 637-56.
4. Moreau, J.; Guillon, E.; Aplincourt, P.; Pierrard, J.-C.; Rimbault, J.; Port, M.; Aplincourt, M., *European Journal of Inorganic Chemistry* **2003**, 2003 (16), 3007-3020.
5. Green, K. N.; Viswanathan, S.; Rojas-Quijano, F.A.; Kovacs, Z.; Sherry, A. D., *Inorganic chemistry* **2011**, 50 (5), 1648-55.
6. BENZAZEPINONE COMPOUND, EP2221308, 2010.
7. Woods, M.; Kiefer, G. E.; Bott, S.; Castillo-Muzquiz, A.; Eshelbrenner, C.; Michaudet, L.; McMillan, K.; Mudigunda, S. D. K.; Ogrin, D.; Tirsoś, G.; Zhang, S.; Zhao, P.; Sherry, A. D., *Journal of the American Chemical Society* **2004**, 126 (30), 9248-9256.

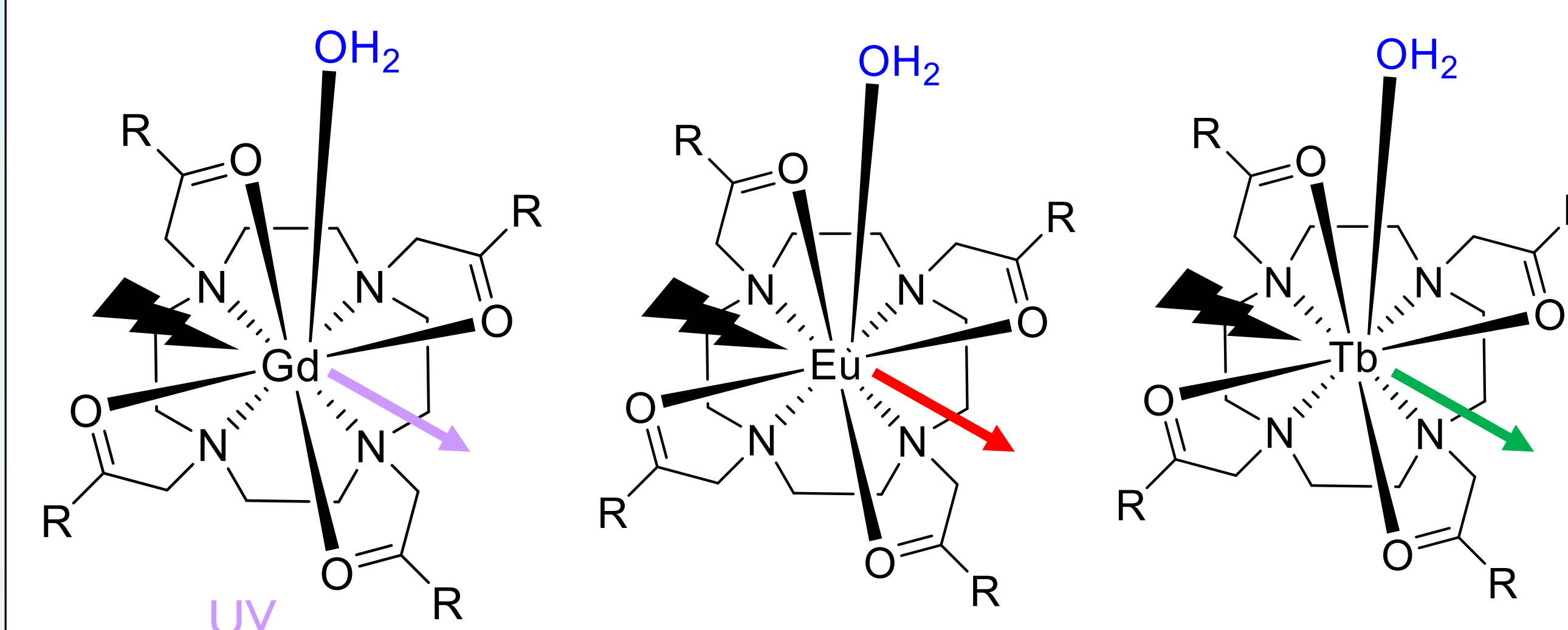
Future Work

Upon successful synthesis of the target compound, it will be incorporated into a hydrogel as a load-bearing cross-linker which will then be mechanically strained.

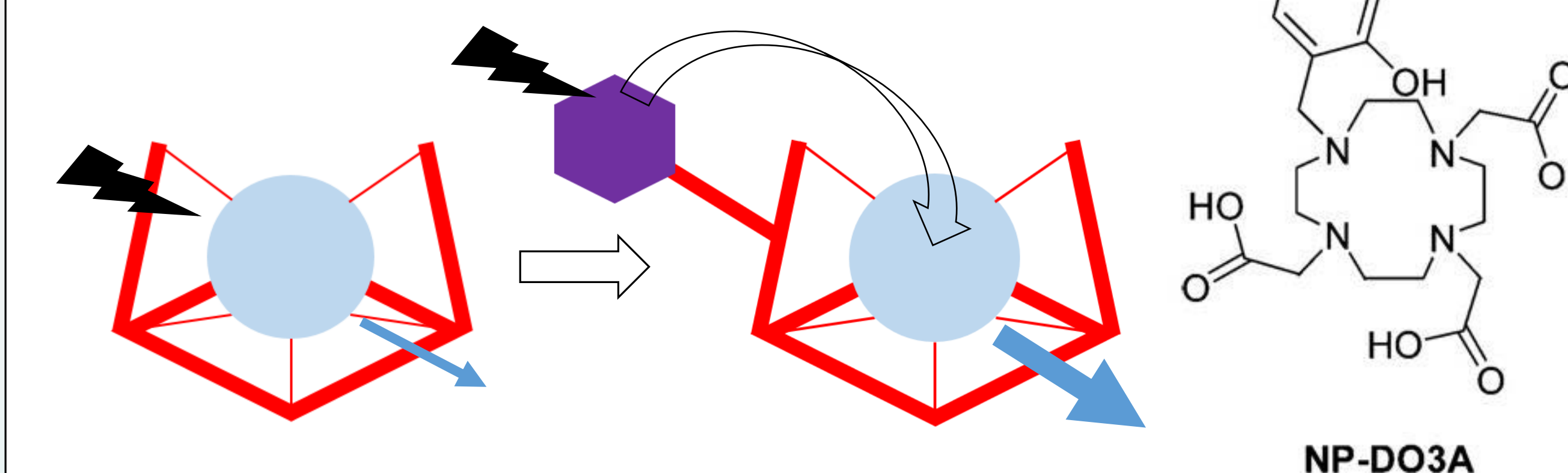


Modifications:

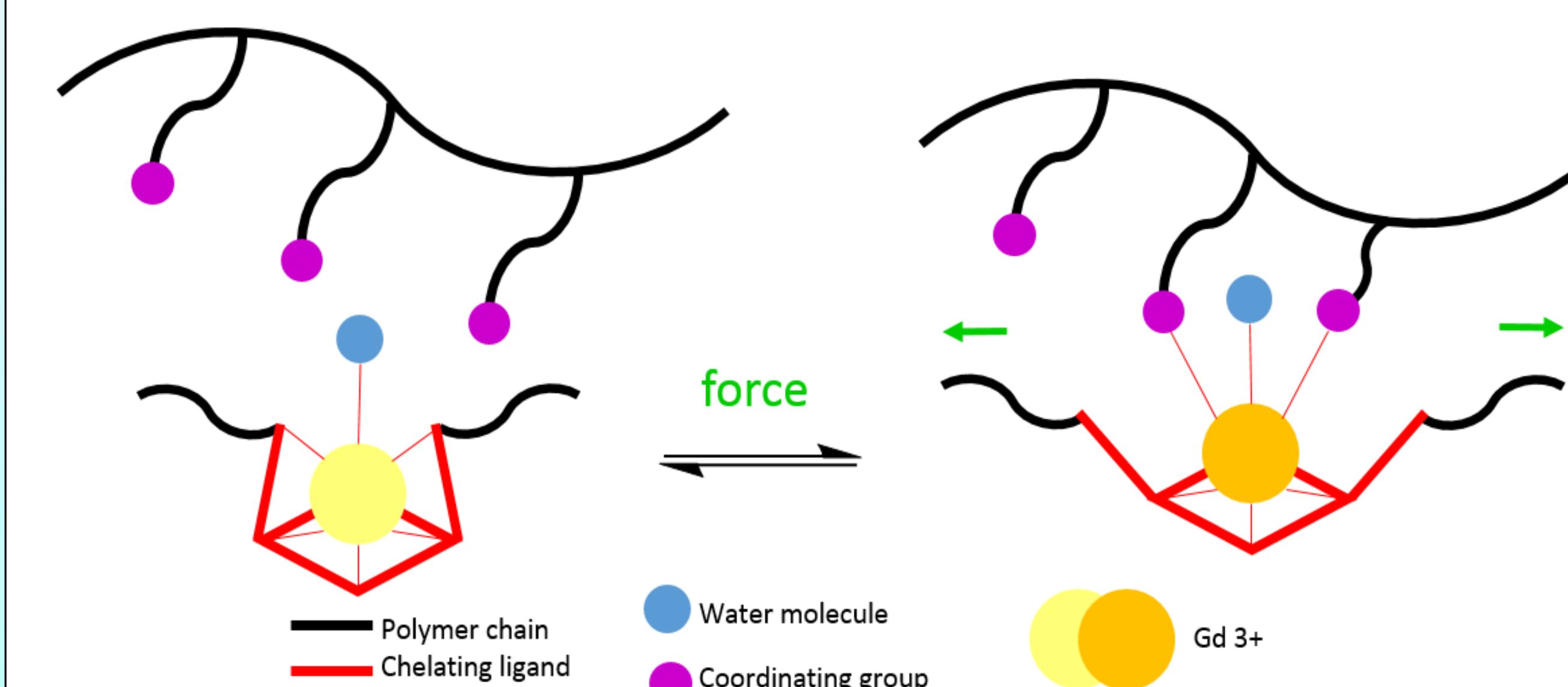
- Different metal⁵; Europium (Eu), Terbium (Tb)



- Sensitizing pendant on ligand⁷



Incorporation of this complex into a different polymer, for example one that had dangling amine groups, could facilitate cross-bridging upon the application of force. The goal of this would be to create a self-strengthening material.



Acknowledgements

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